

EP1213294A1

Publication Title:

Purifying process for phosphatidylserine

Abstract:

Abstract of EP 1213294

(A1) Translate this text The present invention relates to a purifying process for phosphatidylserine, the latter being prepared by trans-phosphatidylation of phosphatidylcholine with serine in presence of the enzyme D-phospholipase and containing as impurities hydrophilic compounds, proteins and inorganic salts.

Courtesy of <http://v3.espacenet.com>

(19)



Europäisches Patentamt

European Patent Office

Office européen des brevets



(11)

EP 1 213 294 A1

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:
12.06.2002 Bulletin 2002/24

(51) Int Cl.7: **C07F 9/10**

(21) Application number: **01124577.6**

(22) Date of filing: **15.10.2001**

(84) Designated Contracting States:
**AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE TR**
Designated Extension States:
AL LT LV MK RO SI

(72) Inventors:
• **De Ferra, Lorenzo**
00162 Rome (IT)
• **Massardo, Pietro**
00154 Rome (IT)

(30) Priority: **05.12.2000 IT MI002631**

(74) Representative: **Gervasi, Gemma, Dr. et al**
Notarbartolo & Gervasi S.p.A.,
Corso di Porta Vittoria, 9
20122 Milano (IT)

(71) Applicant: **CHEMI S.p.A.**
20092 Cinisello Balsamo (Milano) (IT)

(54) **Purifying process for phosphatidylserine**

(57) The present invention relates to a purifying process for phosphatidylserine, the latter being prepared by trans-phosphatidylation of phosphatidylcholine with serine in presence of the enzyme D-phospholipase and containing as impurities hydrophilic compounds, proteins and inorganic salts.

EP 1 213 294 A1

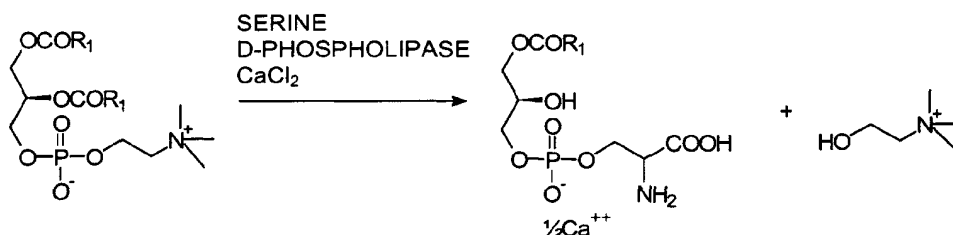
DescriptionFIELD OF THE INVENTION

[0001] The present invention relates to a purifying process for phosphatidylserine, a phospholipid which is useful in particular for the preparation of pharmaceutical compositions used in the treatment of involuntal brain syndromes of various nature, but also in the preparation of particular liposome formulations and of dietetic compositions based on natural lecithins.

STATE OF THE ART

[0002] Phosphatidylserine is a phospholipid widely present in nature; it is one of the main components of cell membranes in animal organisms and is present in particularly large amounts in mammals' brain tissues. Medical literature mentions interesting properties of phosphatidylserine, among which the most significant relates to its effectiveness in improving mnemonic abilities.

[0003] The trans-phosphatidylation of phosphatidylcholine with serine in presence of the enzyme D-phospholipase according to the following scheme is the most convenient reaction for the industrial production of phosphatidylserine.



[0004] The reaction can be carried out both in an aqueous ambient, as described in the EP 1 048 738, and in a system containing beyond an aqueous phase also an organic solvent unmixed with water, preferably toluene; this second method is described in US Patent no. 5,700,668.

[0005] Whatever the method of trans-phosphatidylation used, the phosphatidylserine which can be isolated at the end of the process contains in any case large amounts of hydrophilic impurities, such as serine, choline and their salts which are present in the aqueous phase.

[0006] Another aspect which should be remarked is that, by measuring the enzymatic activity of the type D-phospholipase of the products according to the method described in related literature (*Biotechn. Techn.*, 7, 795 (1993)), every gram of product is found to have about 2 international units of enzymatic activity.

[0007] In compliance with the general request of strict criteria for purity, the removal of said impurities from the product is highly important.

[0008] The methods which are commonly used to remove hydrophilic substances from organic solutions of phospholipids have proved to be ineffective. As a matter of fact, by extracting a toluene solution of the phospholipid mixture obtained as described in US Patent no. 5,700,668, with the same amount of water, and then by separating the phases, it is found that the content of serine in the organic phase and the activity of D-phospholipase are practically unchanged with respect to the values found before the extraction with water.

[0009] Similarly, the elution of the same organic solution on a chromatographic column containing chromatography silica conditioned with toluene, in which the product elution is completed with toluene, results in a very small removal of serine and of the enzymatic activity from the phospholipid.

[0010] In another experiment the organic solution of phosphatidylserine to be purified because of serine is added with acetone: the precipitated product shows, after being analyzed, a ratio of serine to phosphatidylserine substantially unchanged with respect to the starting product.

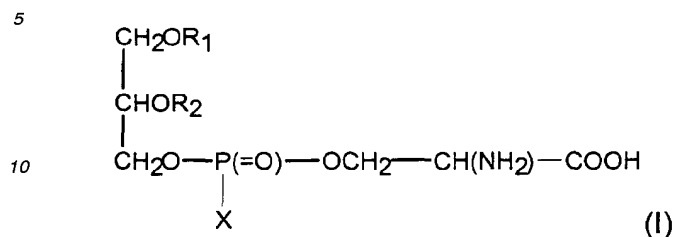
[0011] Therefore, there is always the problem related to the availability of an effective purifying process for phosphatidylserines prepared by trans-phosphatidylation of phosphatidylcholine with serine in presence of the enzyme D-phospholipase, and containing as impurities hydrophilic compounds, proteins and inorganic salts.

SUMMARY OF THE INVENTION

[0012] The Applicant has now surprisingly found that hydrophilic impurities, proteins and inorganic salts can be suc-

cessfully removed from a solution of phosphatidylserine in an organic solvent by extraction with water, provided that a polar organic solvent is added to the system.

[0013] The object of the present invention is therefore a purifying process for phosphatidylserines having formula (I)



where R_1 e R_2 , identical or different, are a C_{10} - C_{30} acyl group; X is OH or OM, where M is chosen from the group of alkali metals, alkaline-earth metals, ammonium and alkyl ammonium, and where the serine portion is in D, L or racemic form, and preferably in L form, comprising the extraction of said phosphatidylserines from a solution in a hydrocarbon solvent with a mixture of water and a polar organic solvent.

DETAILED DESCRIPTION OF THE INVENTION

[0014] In the specific case of phosphatidylserines prepared according to the description contained in US Patent no. 5,700,668 by trans-phosphatidylation in a two-phase system made of an aqueous phase and a toluene phase, the process according to the invention can be carried out directly on the toluene phase after separating the latter from the aqueous phase at the end of the reaction.

[0015] In the case of phosphatidylserines prepared following other methods, the purifying process can be advantageously carried out by stirring the product in a mixture containing a hydrocarbon solvent, water and a polar organic solvent.

[0016] Also in this case the concerned phospholipid is present in the hydrocarbon phase, whereas hydrophilic impurities are to be found in the aqueous phase.

[0017] After separating the phases phosphatidylserine can be isolated using known methods such as precipitation with acetone.

[0018] According to the present process aromatic or aliphatic solvents can be used as hydrocarbon solvent; among aromatic solvents toluene or xylene are preferred; among aliphatic solvents it is preferable to use n-heptane, n-hexane or cyclohexane.

[0019] As polar organic solvent alcoholic solvents can be used, containing for instance 1 to 5 carbon atoms. According to preferred embodiments of the invention, said alcoholic solvent is chosen among secondary and tertiary alcohols; more preferably isopropanol is used.

[0020] The extraction according to the invention can be carried out at temperatures between 0 and 70°C, and preferably between 20 and 30°C.

[0021] The amount of hydrocarbon solvent is between 4 and 30 liters/kg of phospholipid to be purified, and preferably between 6 and 12 liters.

[0022] The volume ratio between water and hydrocarbon solvent is between 0.2 and 5, and preferably between 0.3 and 1. The volume ratio between polar organic solvent and hydrocarbon solvent is between 0.2 and 2, and preferably between 0.3 and 1.2.

[0023] The method described can be applied to phosphatidylserines with different acyclic chains and allows to purify both products deriving from the trans-phosphatidylation of phosphatidylcholines of natural origin such as soybean, rape or egg yolk, and phosphatidylcholines synthesized with fat acids, both saturated such as myristic acid, palmitic acid or stearic acid, and unsaturated such as oleic acid or linoleic acid.

[0024] The following examples merely aim at disclosing the present invention without limiting its object.

EXAMPLE 1

Preparation of phosphatidylserine in a two-phase system

[0025] 400 g of a non de-oiled fraction of soybean lecithin, containing 32% of phosphatidylcholine and 50% of triglycerides, are charged into a flask together with 3 l of toluene. A solution of 22.7 g of calcium chloride, 27.6 g of trihydrated sodium acetate and 62.5 g of L-serine in 1.3 l of a solution of D-phospholipase with an activity of 3 KU/l is separately prepared. The solution is brought to pH = 4.2 with acetic acid. The two solutions are united and kept under

stirring at 25°C for 8 hours. After filtration the phases are separated, one tenth of the toluene phase is concentrated under vacuum (the remaining part is used for the experiments described in examples 2, 3, 4, 5); the residue is added to 500 ml of acetone at room temperature. The product is filtered and dried, thus obtaining 27 g of a mixture of phospholipids with a content of phosphatidylserine of 50%.

[0026] TLC analysis (eluant: mixture of chloroform, methanol, water and 28% aqueous solution of ammonia 300:100:10:12; detector: ninhydrin) shows the presence of 1.5% of L-serine in the final product.

[0027] The presence of D-phospholipase in the product is determined with a method described in related literature (*Biotechn. Techn.*, 7, 795 (1993)); an activity of 2.1 IU/g is found.

EXAMPLE 2

Purifying of phosphatidylserine by extraction with i-propanol and water

[0028] 330 ml of the toluene solution prepared as described in Example 1 are used. This solution is added with 240 ml of i-propanol and 150 ml of water. The whole is stirred at room temperature and settled, then the phases are separated. The toluene phase is concentrated and its residue is added under stirring to 500 ml of acetone.

[0029] The product has a content of L-serine of 0.3%. D-phospholipase is below the determination limit used in the method (0.1 IU/g).

EXAMPLE 3

Purifying of phosphatidylserine by extraction with i-propanol and water in different ratios

[0030] 330 ml of the toluene solution prepared as described in Example 1 are used. This solution is added with 150 ml of isopropanol and 240 ml of water. The whole is stirred at room temperature and settled, then the phases are separated. The toluene phase is concentrated and its residue is added under stirring to 500 ml of acetone.

[0031] The product has a content of L-serine of 0.2%. D-phospholipase is below the determination limit used in the method (0.1 IU/g).

EXAMPLE 4

Purifying of phosphatidylserine by extraction with ethanol and water

[0032] 330 ml of the toluene solution prepared as described in Example, 1 are used. This solution is added with 150 ml of ethanol and 150 ml of water. The whole is stirred at room temperature and settled, then the phases are separated. The toluene phase is concentrated and its residue is added under stirring to 500 ml of acetone.

[0033] The product has a content of L-serine of 0.2%. D-phospholipase is below the determination limit used in the method (0.1 IU/g).

EXAMPLE 5

Preparation of phosphatidylserine in a two-phase system water/n-heptane and purifying with i-propanol and water

[0034] 40 g of the same type of soybean lecithin used as raw material in Example 1 are charged into a flask together with 300 ml of n-heptane. A solution of 2.3 g of calcium chloride, 2.8 g of trihydrated sodium acetate and 6.3 g of L-serine in 0.13 l of a solution of D-phospholipase with an activity of 3 KU/l is separately prepared. The solution is brought to pH = 4.2 with acetic acid. The two solutions are joined and kept under stirring at 25°C for 24 hours. After filtration the phases are separated; the heptane phase is added with 150 ml of i-propanol and 150 ml of water. After stirring and settling the two phases are separated. The heptane phase is concentrated at low pressure. The residue is added to 500 ml of acetone at room temperature. The product is filtered and dried, thus obtaining 26.5 g of a mixture of phospholipids with a content of phosphatidylserine of 48%.

[0035] The product has a content of L-serine of 0.2%. D-phospholipase is below the determination limit of the method (0.1 IU/g).

EXAMPLE 6

Preparation of phosphatidylserine by trans-phosphatidylation in absence of organic solvent

[0036] 40 g of a mixture of phospholipids obtained by extraction with ethyl alcohol of soybean lecithin, whose main component is phosphatidylcholine (65%), are charged into a separatory funnel together with a solution prepared by dissolving 60 g of calcium chloride in 800 ml of water; the mixture is kept under stirring at 25°C for an hour. After 4 hours at rest 700 ml of aqueous phase are unloaded from the bottom tap. A solution containing 64 g of L-serine in 130 ml of 0.1 M acetate buffer at pH 4.5 is separately prepared. This solution is added with 300 mg of D-phospholipase in lyophilized form with an activity of 1.0 IU/mg. The resulting solution is added to the aqueous dispersion of phospholipids previously prepared. The mixture is kept under stirring at 45°C for 3 hours. The reaction product is separated by filtration; after being washed with water it weighs 60 g.

[0037] An aliquot is dried to determine serine and D-phospholipase. The result is that the content of serine is of 7% by weight and the activity of D-phospholipase is of 2 IU/g.

EXAMPLE 7

Purifying of phosphatidylserine as obtained in Example 6

[0038] 5 g aliquots of the moist product obtained as described in Example 6 are used for purifying tests by dissolution in a mixture of solvents containing water, a hydrocarbon solvent (solvent 1) and an alcohol solvent (solvent 2).

[0039] In all tests phosphatidylserine is present, after stirring and settling, only in the hydrocarbon phase. The hydrocarbon phase is concentrated and precipitated from acetone. The dried products are analyzed by measuring the content of serine and the activity of D-phospholipase. The following Table 1 shows the purifying conditions used for phosphatidylserine and the amounts of serine and D-phospholipase which are present in the final product.

Table 1 -

Purifying of phosphatidylserine						
Test	1	2	3	4	5	6
Solvent 1	n-heptane	n-heptane	n-heptane	n-heptane	n-heptane	toluene
Amount of solvent 1 (ml)	50	50	50	50	50	50
Solvent 2	methanol	i-propanol	i-propanol	i-propanol	i-propanol	i-propanol
Amount of solvent 2 (ml)	60	15	15	30	30	50
Amount of water (ml)	25	15	30	15	30	40
Serine in final product (%)	1.0	0.8	0.2	1.3	0.3	0.6
D-phospholipase in final product (IU/g)	<0.1	<0.1	<0.1	0.1	<0.1	<0.1

EXAMPLE 8

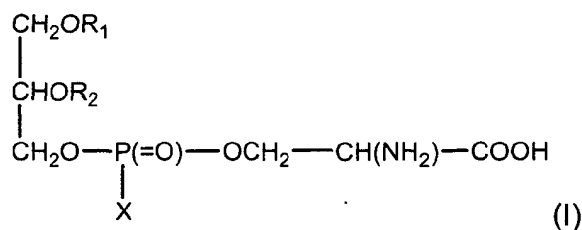
Preparation and purifying of dioleoylphosphatidylserine (DOPS)

[0040] 40 g of dioleoylphosphatidylcholine (DOPC) are charged into a flask together with 0.3 l of toluene. A solution of 2.3 g of calcium chloride, 2.8 g of trihydrated sodium acetate and 6.3 g of L-serine in 0.13 l of a solution of D-phospholipase with an activity of 3 KU/l is separately prepared. The solution is brought to pH = 4.2 with acetic acid. The two solutions are joined and kept under stirring at 25°C for 8 hours. The phases are then separated, the upper phase is added with 240 ml of i-propanol and 150 ml of water. The whole is stirred at room temperature and settled, then the phases are separated. The toluene phase is concentrated and its residue is added under stirring to 500 ml of acetone. The solid is filtered and dried under vacuum, thus obtaining 39 g of DOPS as calcium salt.

[0041] The product has a content of L-serine of 0.2%. D-phospholipase is below the determination limit used in the method (0.1 IU/g).

Claims

1. Purifying process for phosphatidylserine having formula (I)



where R_1 e R_2 , identical or different, are a C_{10} - C_{30} acyl group; X is OH or OM, where M is chosen from the group of alkali metals, alkaline-earth metals, ammonium and alkyl ammonium, and where the serine portion is in D, L or racemic form, and preferably in L form, comprising the extraction of said phosphatidylserines from a solution in a hydrocarbon solvent with a mixture of water and a polar organic solvent.

2. Process according to claim 1, in which said hydrocarbon solvent is chosen among toluene, xylene, n-heptane, n-hexane or cyclohexane.
3. Process according to claim 1, in which said hydrocarbon solvent is used in an amount between 4 and 30 liters/kg of phosphatidylserine to be purified.
4. Process according to claim 3, in which said hydrocarbon solvent is used in an amount between 6 and 12 liters/kg of phosphatidylserine to be purified.
5. Process according to claim 1, in which said polar organic solvent is an alcohol solvent.
6. Process according to claim 5, in which said alcohol solvent is chosen among alcohols containing 1 to 5 carbon atoms.
7. Process according to claim 5, in which said alcohol solvent is chosen among secondary and tertiary alcohols.
8. Process according to claim 5, in which said alcohol solvent is i-propanol.
9. Process according to claim 1, in which said polar organic solvent is used in an amount between 0.2 and 2 liters/kg of hydrocarbon solvent used.
10. Process according to claim 9, in which said polar organic solvent is used in an amount between 0.3 and 1.2 liters/kg of hydrocarbon solvent used.
11. Process according to claim 1, in which the amount of water used is between 0.2 and 5 liters/kg of hydrocarbon solvent used.
12. Process according to claim 11, in which the amount of water used is between 0.3 and 1 liter/kg of hydrocarbon solvent used.
13. Process according to claim 1, in which said extraction is carried out at a temperature between 0 and 70°C.
14. Process according to claim 13, in which said extraction is carried out at a temperature between 20 and 30°C.
15. Process according to claim 1, in which said phosphatidylserines having formula (I) are prepared by trans-phosphatidylation of phosphatidylcholines of natural or synthetic origin.



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 01 12 4577

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
D, A	EP 1 048 738 A (CHEMI S.P.A.) 2 November 2000 (2000-11-02) * the whole document *	1-15	C07F9/10
D, A	EP 0 922 707 A (CHEMI S.P.A.) 16 June 1999 (1999-06-16) * the whole document *	1-15	
A	US 5 084 215 A (JOHN J. KEARNS) 28 January 1992 (1992-01-28) * the whole document *	1-15	
A	DE 199 17 249 A (LUCAS MEYER GMBH & CO.) 7 September 2000 (2000-09-07) * the whole document *	1-15	
The present search report has been drawn up for all claims			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			C07F
Place of search	Date of completion of the search	Examiner	
THE HAGUE	8 March 2002	Beslier, L	
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document			

EPC FORM 1503 03/82 (P04/01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 01 12 4577

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

08-03-2002

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
EP 1048738	A	02-11-2000	IT	MI990895 A1	30-10-2000
			EP	1048738 A1	02-11-2000
			JP	2000333689 A	05-12-2000
EP 922707	A	16-06-1999	US	5700668 A	23-12-1997
			EP	0922707 A1	16-06-1999
			AT	203057 T	15-07-2001
			AT	208399 T	15-11-2001
			CA	2192257 A1	09-06-1997
			DE	69613801 D1	16-08-2001
			DE	69613801 T2	22-11-2001
			DE	69616834 D1	13-12-2001
			DE	776976 T1	19-02-1998
			EP	0776976 A2	04-06-1997
			ES	2105999 T1	01-11-1997
			IT	MI961686 A1	02-02-1998
			JP	9173092 A	08-07-1997
US 5084215	A	28-01-1992	US	4814111 A	21-03-1989
			US	4714571 A	22-12-1987
DE 19917249	A	07-09-2000	DE	19917249 A1	07-09-2000